

International Workshop  
November 13-14 2010

# What's the colour of soil

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# Introduction

- **Colour** of an object is due to the light leaving its surface. This light depends on:
  - the spectrum of the incident illumination;
  - the reflectance properties of the object surfaces;
  - -angles of illumination and viewing

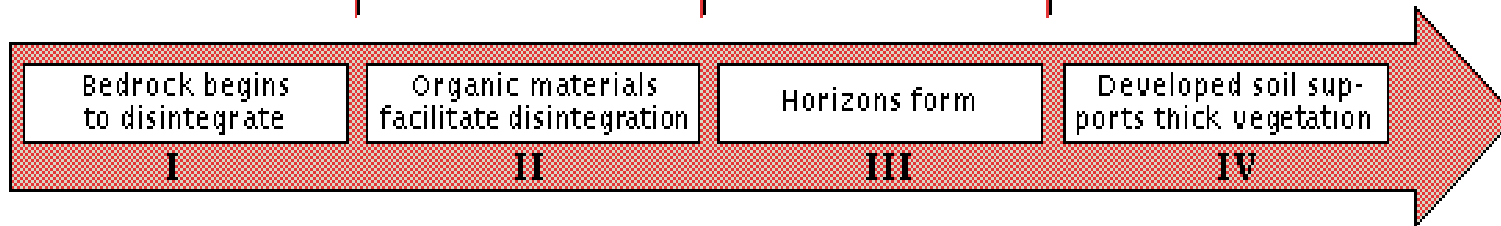
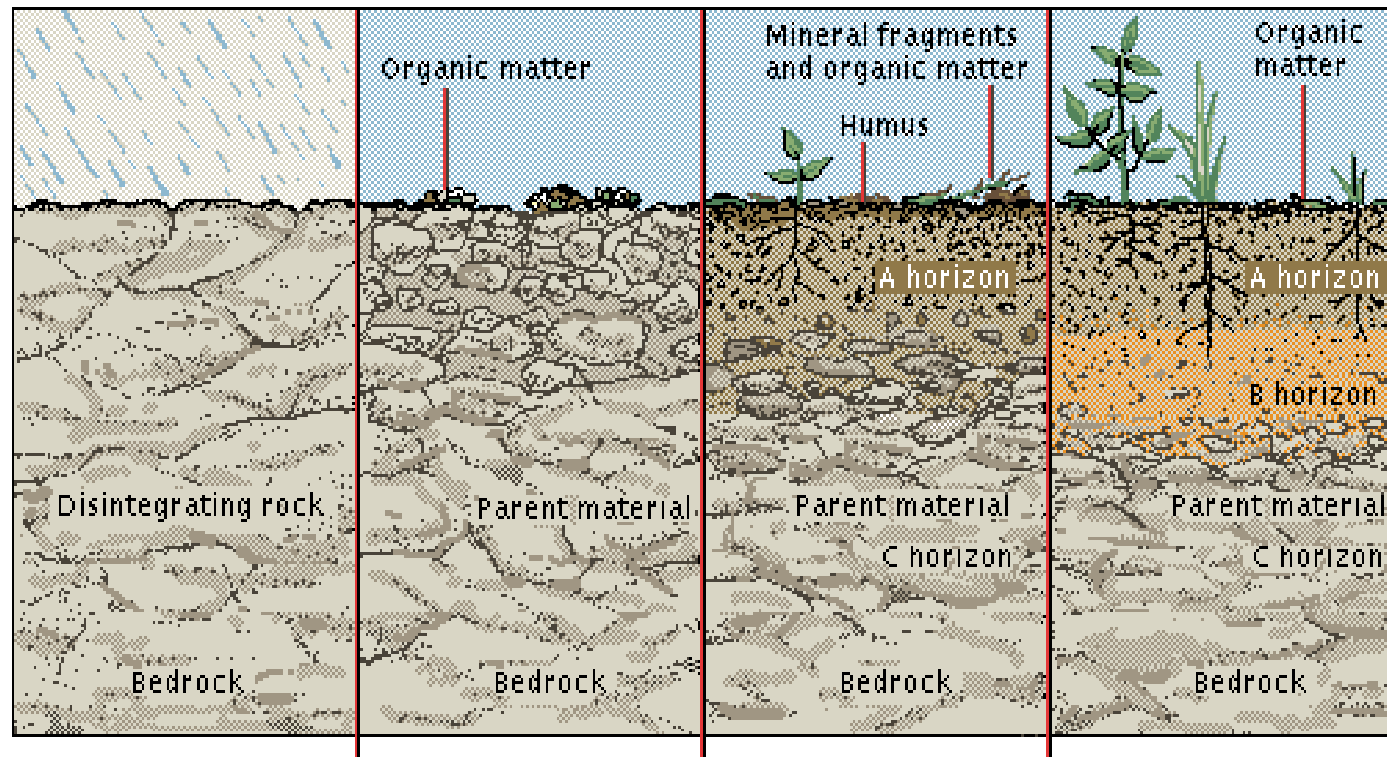
# Introduction

- The incident radiation is partially absorbed by the surface of the solid.
- Absorptions of light arise as result of interaction of light and the valence electrons in the compound.
- The color of the solid is the complement of that of the light absorbed.

# SOIL



# SOIL FORMATION



# SOIL AS BUILDING MATERIAL

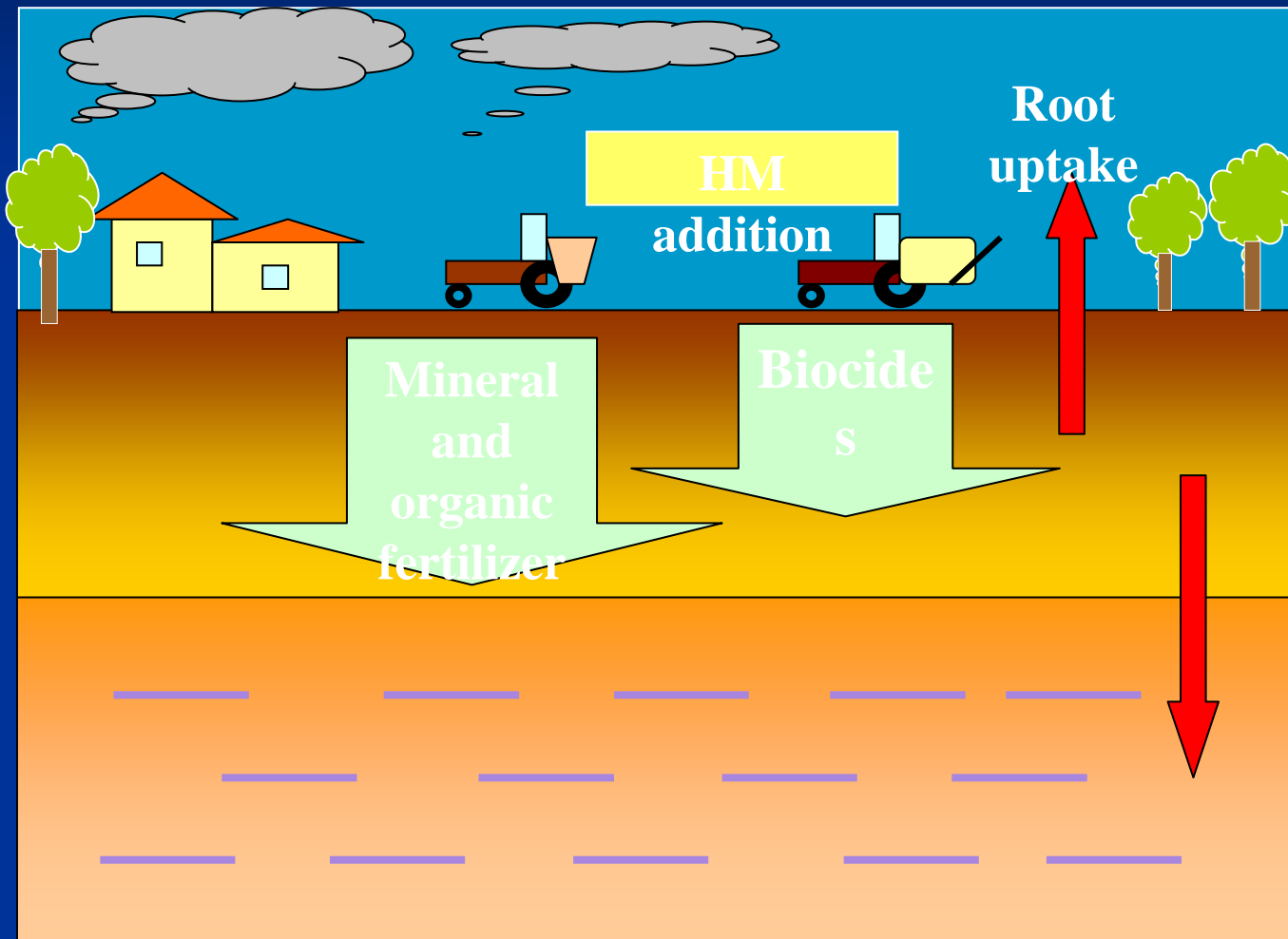




# SOIL AS RECYCLE LABORATORY

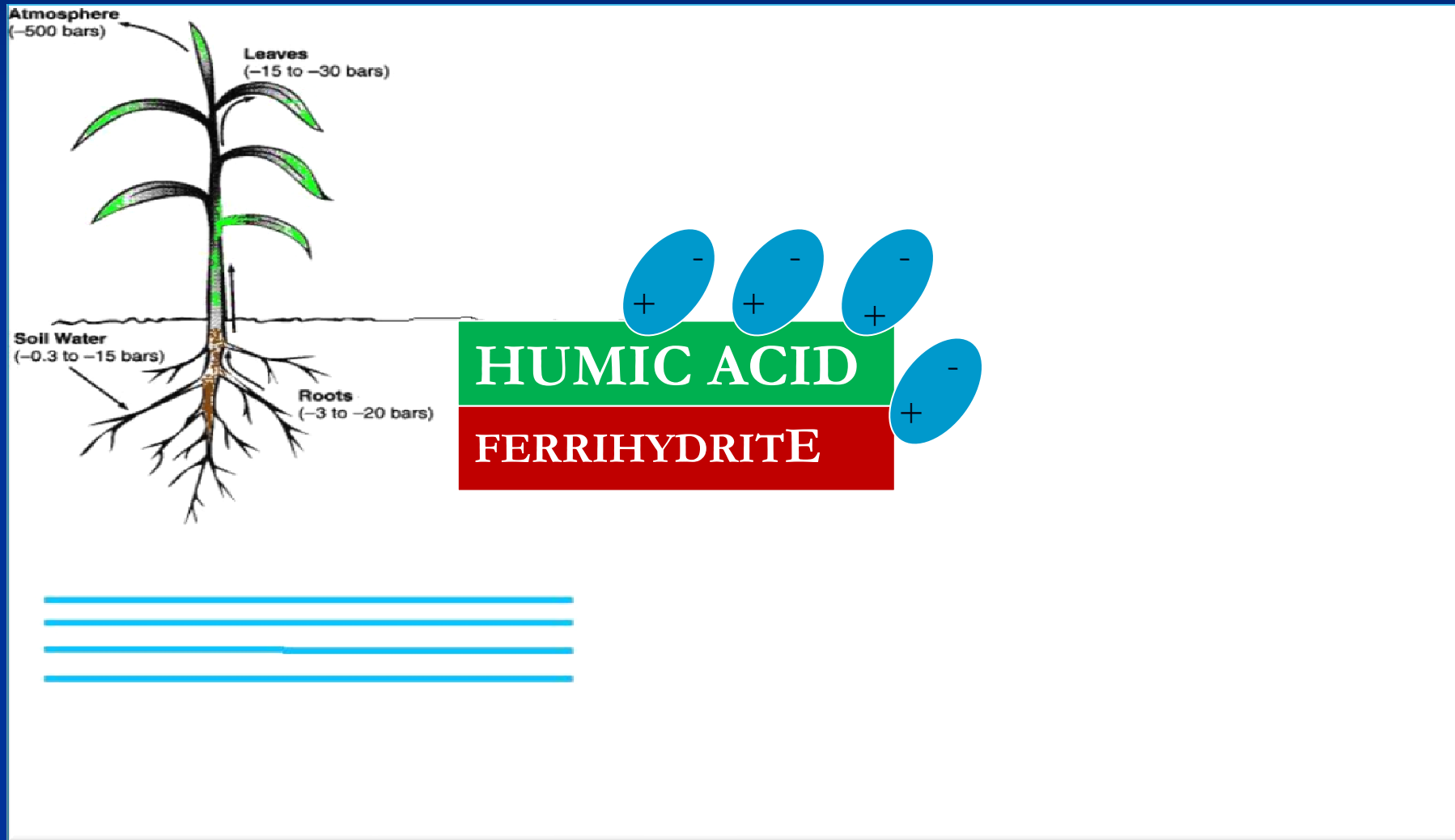


# Soil as filter and nutrient deposit





# XENOBIOTIC RETENTION



# SOIL FOR FARM PRODUCTIONS

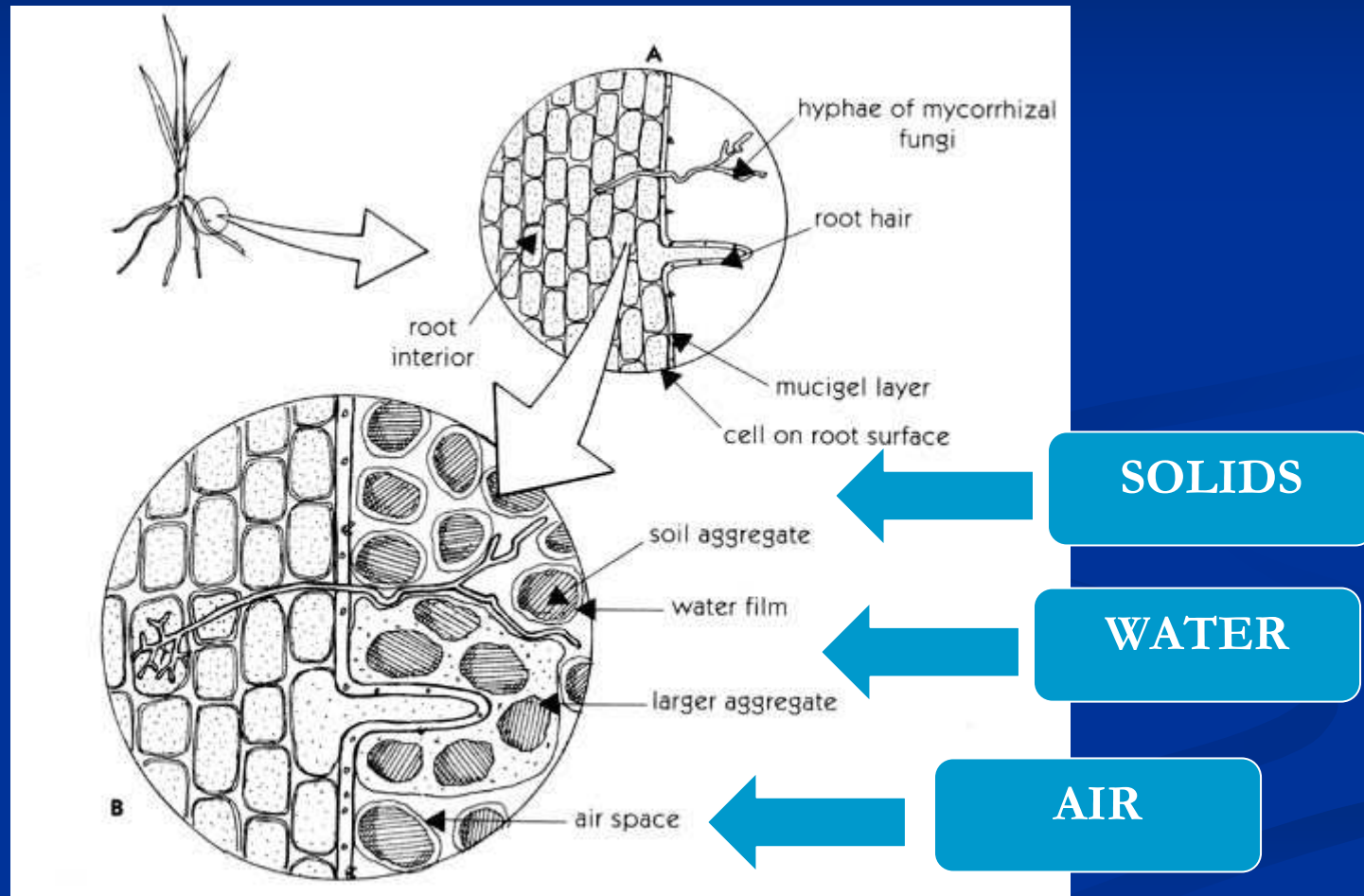


# SOIL AS NUTRIENT SOURCE

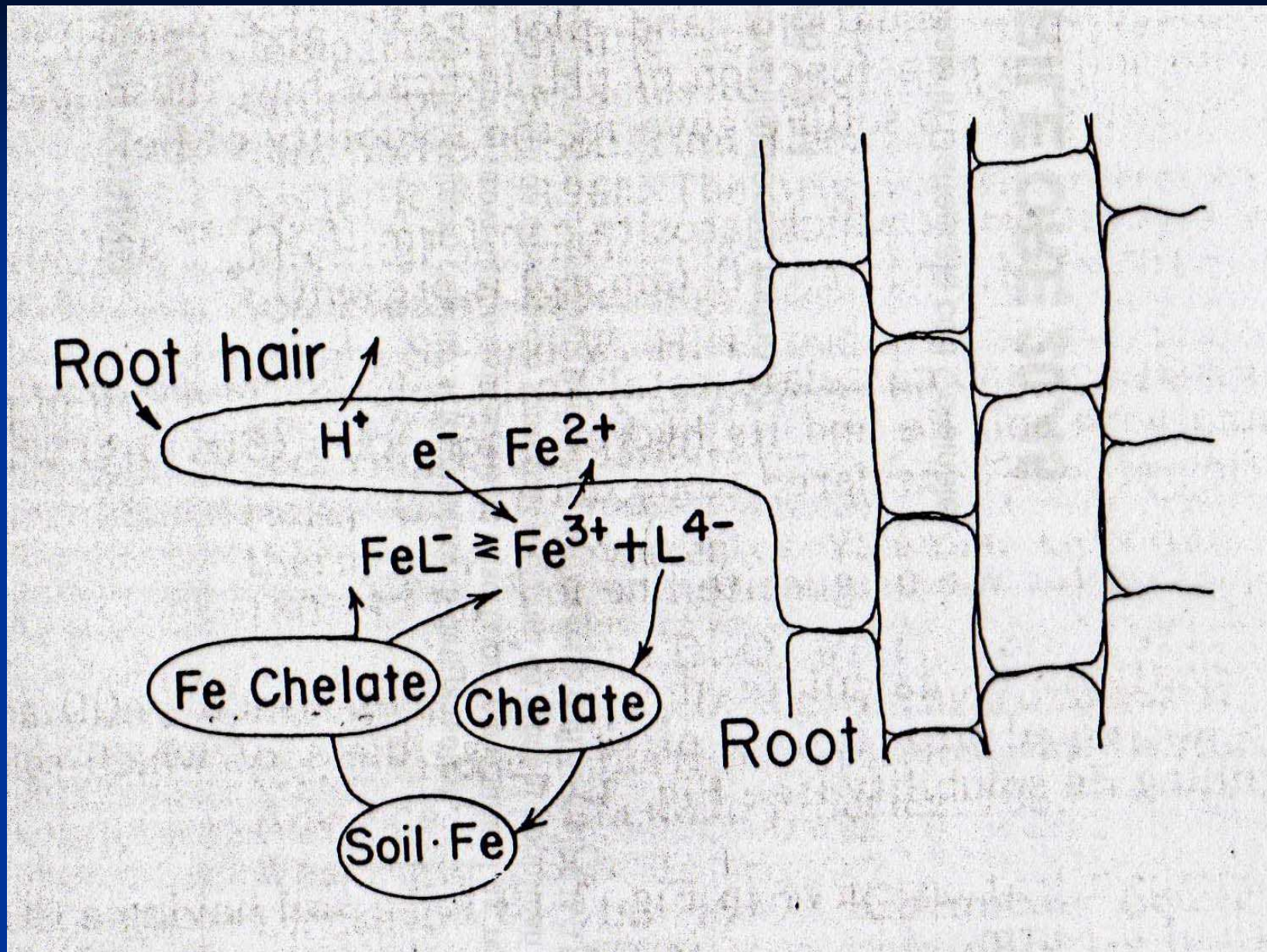
The Periodic Table of the Elements

1 <b>H</b> Hydrogen 1.00794																	2 <b>He</b> Helium 4.003														
3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012182											5 <b>B</b> Boron 10.811	6 <b>C</b> Carbon 12.0107	7 <b>N</b> Nitrogen 14.00674	8 <b>O</b> Oxygen 15.9994	9 <b>F</b> Fluorine 18.9984032	10 <b>Ne</b> Neon 20.1797														
11 <b>Na</b> Sodium 22.989770	12 <b>Mg</b> Magnesium 24.3050											13 <b>Al</b> Aluminum 26.981538	14 <b>Si</b> Silicon 28.0855	15 <b>P</b> Phosphorus 30.973761	16 <b>S</b> Sulfur 32.066	17 <b>Cl</b> Chlorine 35.4527	18 <b>Ar</b> Argon 39.948														
19 <b>K</b> Potassium 39.0983	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.955910	22 <b>Ti</b> Titanium 47.867	23 <b>V</b> Vanadium 50.9415	24 <b>Cr</b> Chromium 51.9961	25 <b>Mn</b> Manganese 54.938049	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933200	28 <b>Ni</b> Nickel 58.6934	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.39	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.61	33 <b>As</b> Arsenic 74.92160	34 <b>Se</b> Selenium 78.96	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.80														
37 <b>Rb</b> Rubidium 85.4678	38 <b>Sr</b> Strontium 87.62	39 <b>Y</b> Yttrium 88.90585	40 <b>Zr</b> Zirconium 91.224	41 <b>Nb</b> Niobium 92.90638	42 <b>Mo</b> Molybdenum 95.94	43 <b>Tc</b> Technetium (98)	44 <b>Ru</b> Ruthenium 101.07	45 <b>Rh</b> Rhodium 102.90550	46 <b>Pd</b> Palladium 106.42	47 <b>Ag</b> Silver 107.8682	48 <b>Cd</b> Cadmium 112.411	49 <b>In</b> Indium 114.818	50 <b>Sn</b> Tin 118.710	51 <b>Sb</b> Antimony 121.760	52 <b>Te</b> Tellurium 127.60	53 <b>I</b> Iodine 126.90447	54 <b>Xe</b> Xenon 131.29														
55 <b>Cs</b> Cesium 132.90545	56 <b>Ba</b> Barium 137.327	57 <b>La</b> Lanthanum 138.9055	72 <b>Hf</b> Hafnium 178.49	73 <b>Ta</b> Tantalum 180.9479	74 <b>W</b> Tungsten 183.84	75 <b>Re</b> Rhenium 186.207	76 <b>Os</b> Osmium 190.23	77 <b>Ir</b> Iridium 192.217	78 <b>Pt</b> Platinum 195.078	79 <b>Au</b> Gold 196.96655	80 <b>Hg</b> Mercury 200.59	81 <b>Tl</b> Thallium 204.3833	82 <b>Pb</b> Lead 207.2	83 <b>Bi</b> Bismuth 208.98038	84 <b>Po</b> Polonium (209)	85 <b>At</b> Astatine (210)	86 <b>Rn</b> Radon (222)														
87 <b>Fr</b> Francium (223)	88 <b>Ra</b> Radium (226)	89 <b>Ac</b> Actinium (227)	104 <b>Rf</b> Rutherfordium (261)	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (263)	107 <b>Bh</b> Bohrium (262)	108 <b>Hs</b> Hassium (265)	109 <b>Mt</b> Meitnerium (266)	110 (269)	111 (272)	112 (277)	113	114																		
																		58 <b>Ce</b> Cerium 140.116	59 <b>Pr</b> Praseodymium 140.90765	60 <b>Nd</b> Neodymium 144.24	61 <b>Pm</b> Promethium (145)	62 <b>Sm</b> Samarium 150.36	63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.92534	66 <b>Dy</b> Dysprosium 162.50	67 <b>Ho</b> Holmium 164.93032	68 <b>Er</b> Erbium 167.26	69 <b>Tm</b> Thulium 168.93421	70 <b>Yb</b> Ytterbium 173.04	71 <b>Lu</b> Lutetium 174.967
																		90 <b>Th</b> Thorium 232.0381	91 <b>Pa</b> Protactinium 231.03588	92 <b>U</b> Uranium 238.0289	93 <b>Np</b> Neptunium (237)	94 <b>Pu</b> Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (262)

# SOIL AS NUTRIENT SOURCE FOR PLANT AND OTHER ORGANISMS









# Soil colour

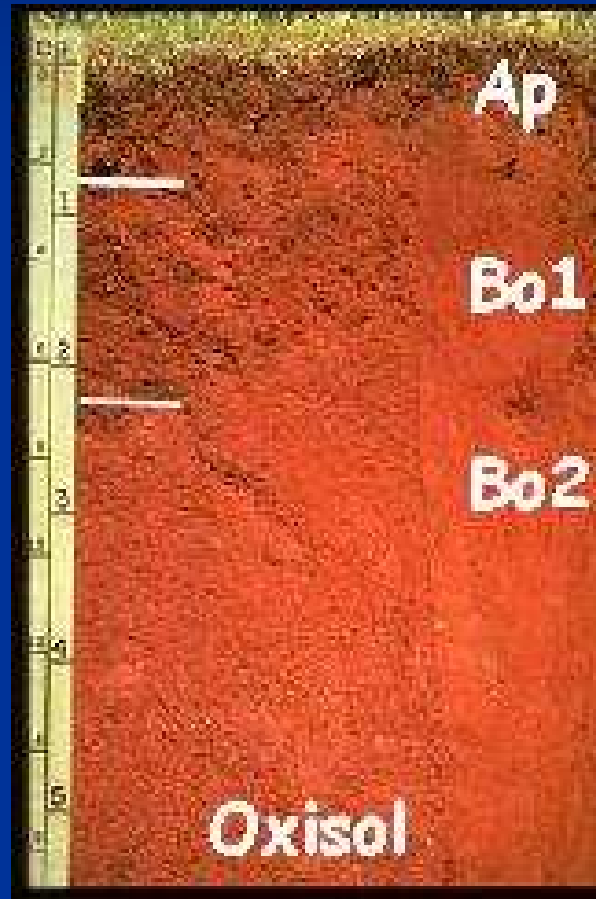




# SOIL PROFILES



# Soil profile





# Soil components and colour

## ■ Organic



# Soil components and colour

Organics





# Soil components and colour

## Inorganics (minerals)

- **Iron oxides**

- Yellow → goethite  $\text{FeOOH}$

- Red → hematite  $\text{Fe}_2\text{O}_3$

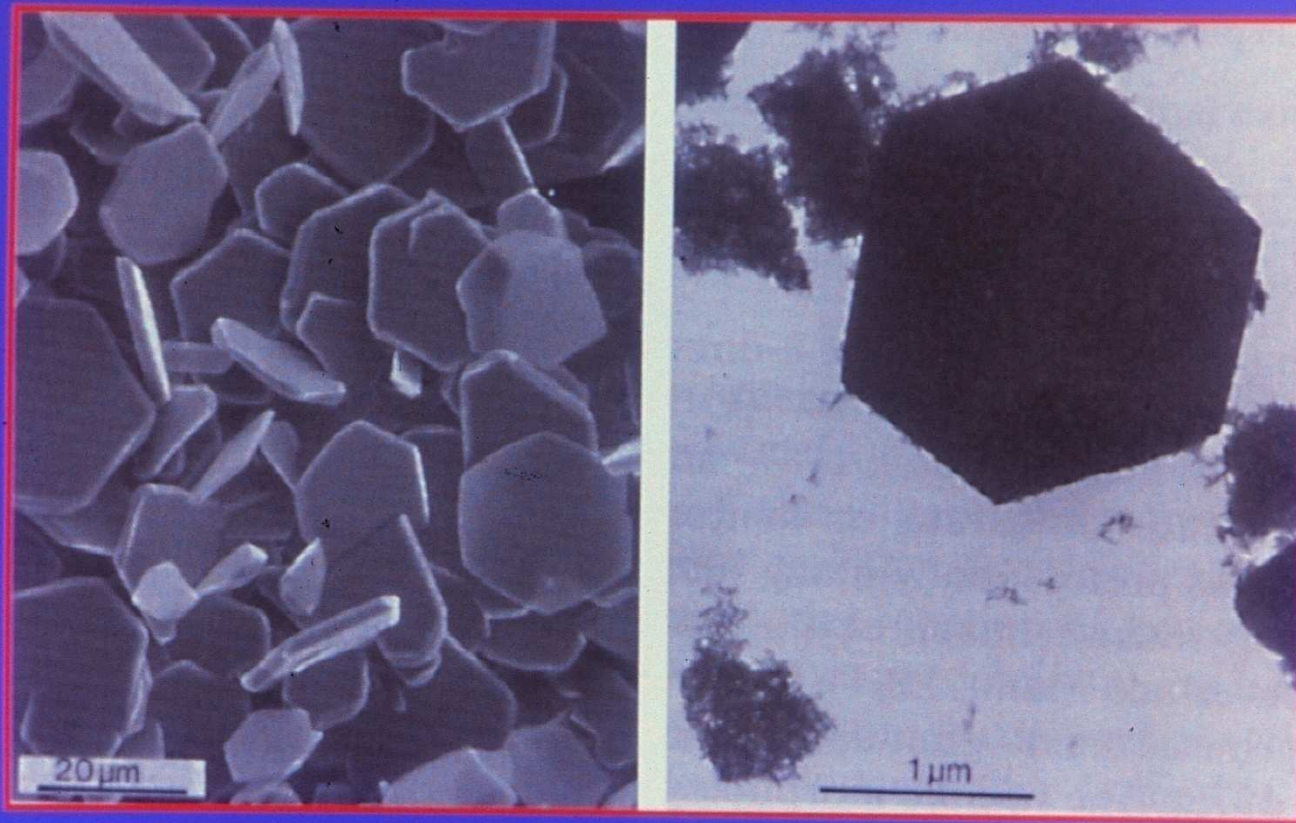
- Brown → Ferrihydrite

- Dark brown → Magnetite,  $\text{Fe}_3\text{O}_4$

- **Manganese oxides** (black colours)

- **Carbonate and sulfate** (white and gray colours)

# Hematite crystals



# Hematite crystal

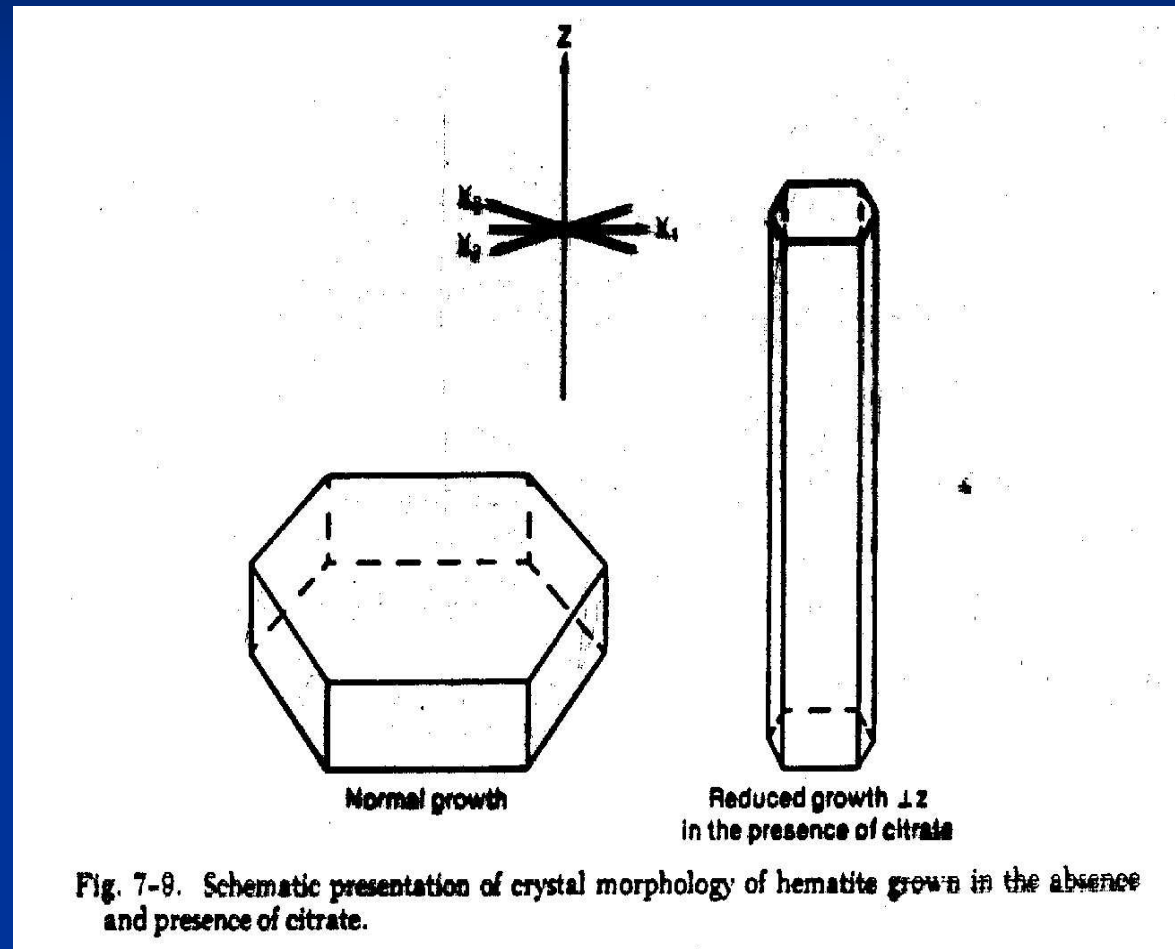


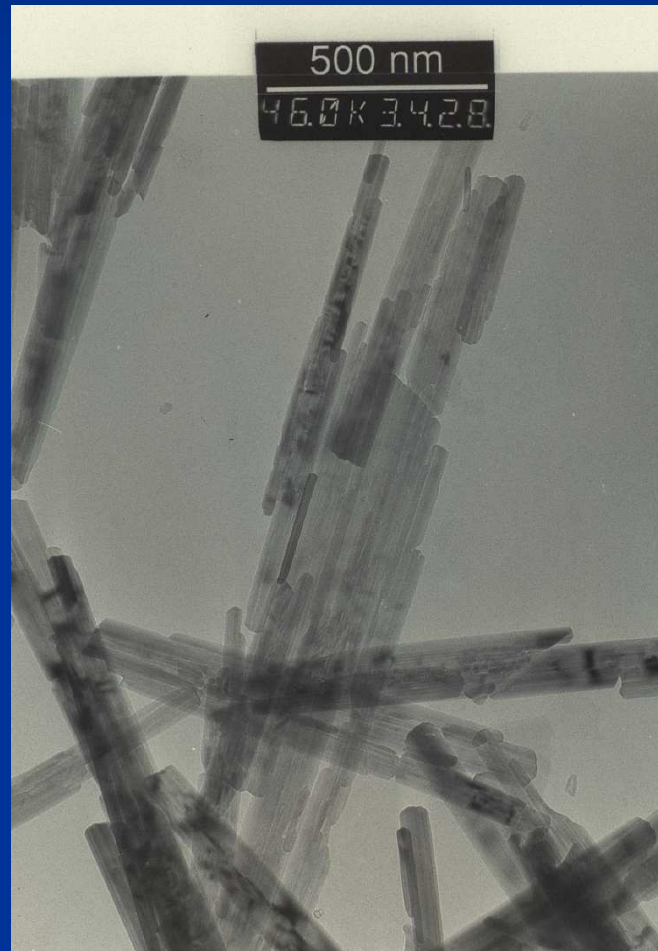
Fig. 7-9. Schematic presentation of crystal morphology of hematite grown in the absence and presence of citrate.

# Red soil, haematic soil



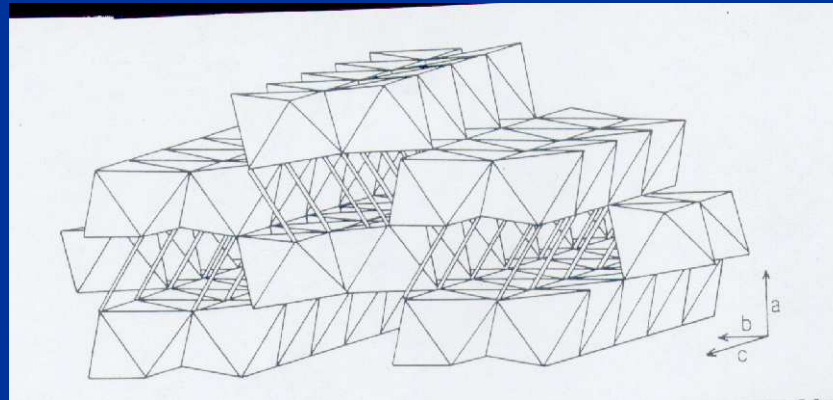


# Goethite crystals, ( $\alpha\text{FeOOH}$ )

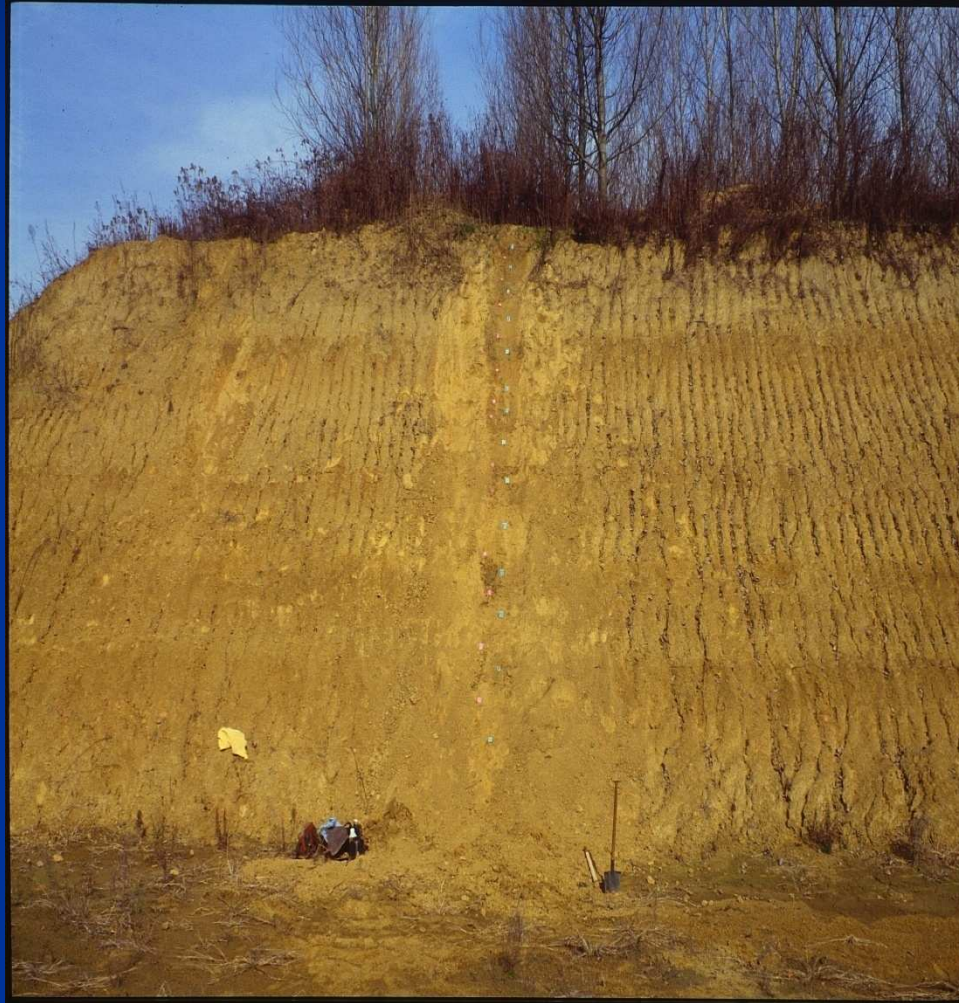




# Goethite structure

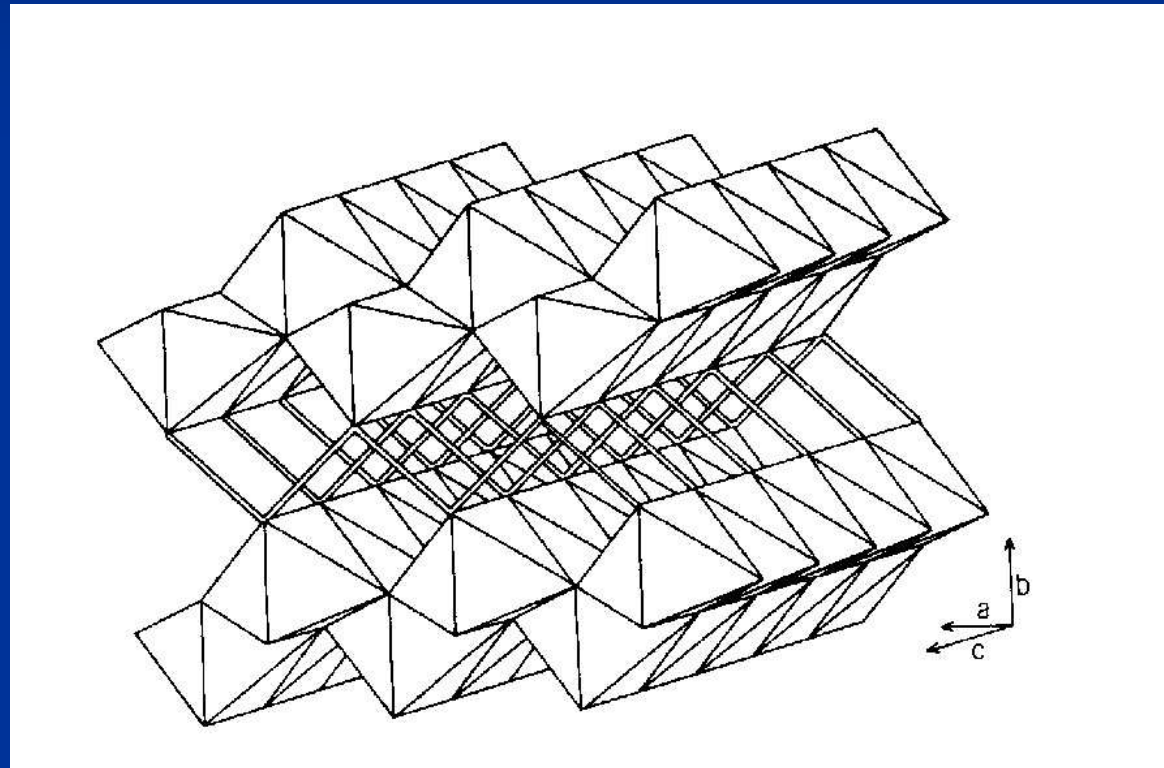


# Yellow soil, Goethitic soil



# Crystal colour

- Cation substitution



# Crystal colour

- Crystal dimension





# Munsell colour chart



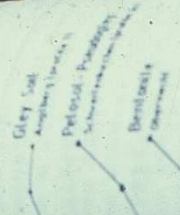
# Colour distribution





III. Clay Minerals

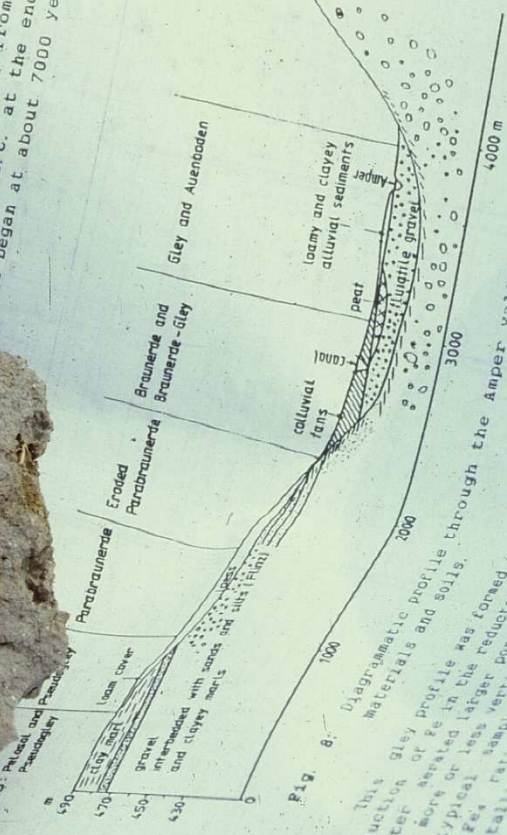
Ver.	X
45	12
66	9
39	12



**Gley**

km wide and has a steep  
 Fig. 8). A similar asymmet-  
 served in the numerous  
 m MNK to SSE. AWAY FROM  
 7,000 years B.C. at the end of  
 the ice began at about 7000 years

**SOILS:**



**Fig. 8:** Diagrammatic profile through the Amper valley showing parent materials and soils.  
 This gley profile was formed in clayey alluvium of the Amper valley. Reduction of iron and manganese in the clayey matrix and its better crystallization led to the formation of a dense, massive, and brittle soil. A typical less layer reductomorphic Fe/Fe<sub>2</sub>O<sub>3</sub> ratio of a dense, massive, and brittle soil. The root system is shallow and the soil is very fertile.

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# Yellow ped

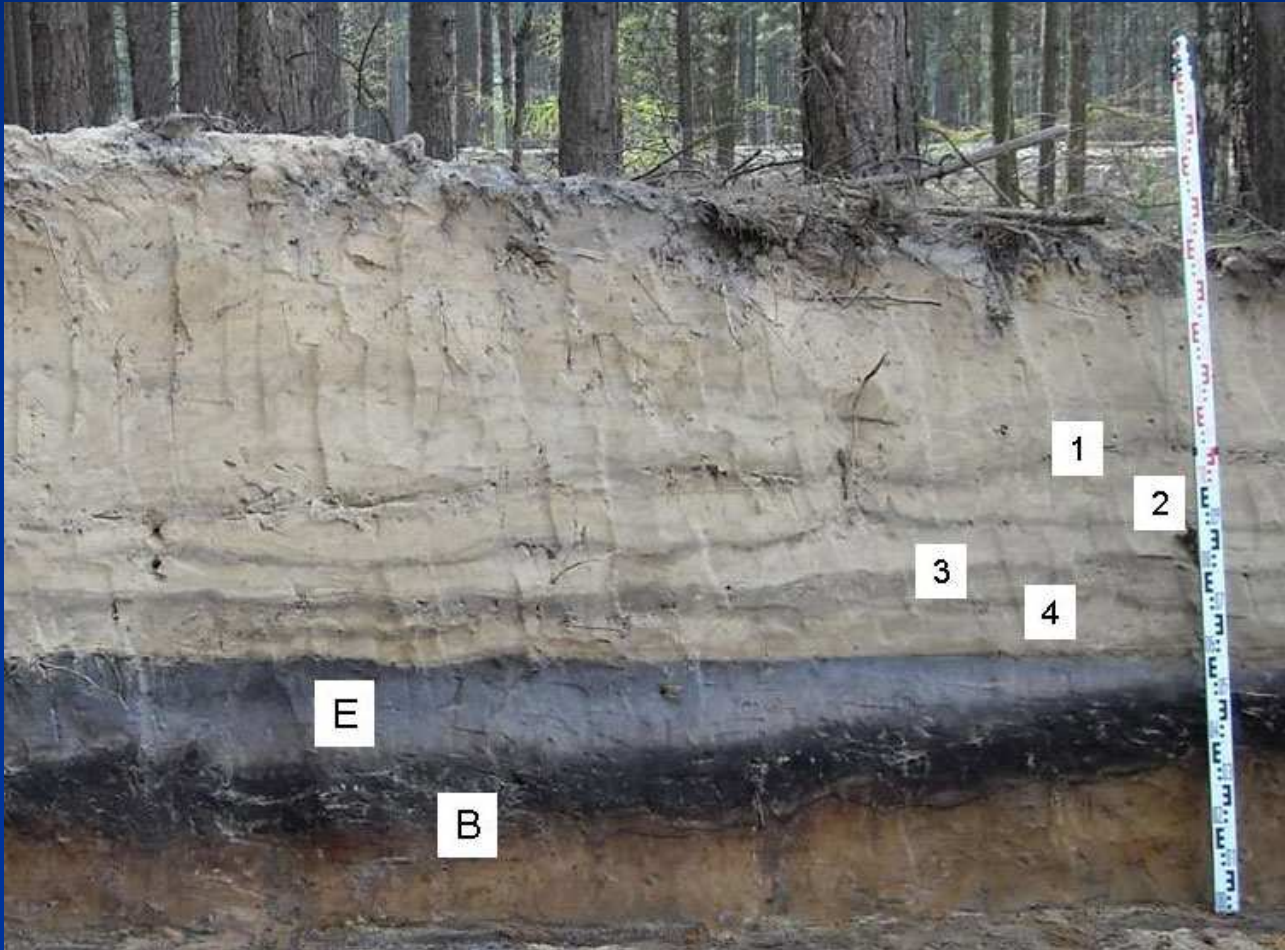






# Podsol







# Podzol, *Great Britain*





# Tasmania, 2009



**Thanks a lot  
for your attention**

